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covers the basics that you may have learned in your chemistry class.

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Section Review 2-4 1.

protons; neutrons 2.

electrons 3. neutrons

4. electrons 5. ionic 6.

The two main types of

chemical bonds are

ionic and covalent

bonds.

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Periodicity; 18.2

Occurrence and

Preparation of the

Representative Metals;

18.3 Structure and

General Properties of

the Metalloids; 18.4

Structure and General

Properties of the

Nonmetals; 18.5

Occurrence,

Preparation, and

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Hydrogen; 18.6
Occurrence,
Preparation, and
Properties of
Carbonates; 18.7
Occurrence,
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Physiology I course at

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isotopes, ion formation,
chemical bonds in ionic
and covalent
compounds, the types
of chemical reactions,
and the naming of
compounds.

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Chapter 2: Introduction

to the Chemistry of Life

Figure 2.1 Foods such

as bread, fruit, and

cheese are rich sources

of biological

macromolecules. The

elements carbon,

hydrogen, nitrogen,

oxygen, sulfur, and

phosphorus are the key

building blocks of the

chemicals found in

living things.

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(Part 1.3)

Biology Chapter 2- The
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Essential Question:

What are the basic
chemical principles
that affect living
things?

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