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For questions 11 - 12, show your work and circle your final answer. 11. If 115 g  $\text{KNO}_3$  are added to 100 g of water at  $35^\circ\text{C}$ , how many grams do not dissolve? 12. What mass of  $\text{KCl}$  would be needed to form a saturated solution if the  $\text{KCl}$  was dissolved in 200 g of water at  $80^\circ\text{C}$ ? !52 g  $\text{KCl}$  100 g  $\text{H}_2\text{O}$  = !104 g  $\text{KCl}$

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200 g H<sub>2</sub>O supersaturated ...

**Use the provided solubility graph to answer the following**

...

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## **Chemistry A Study Of Matter Unit 6 Answer Key**

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