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uses examples to frame the importance of the material. Each topic begins with a motivational example that is investigated in context to that topic. This framing of the material is helpful to all readers, particularly to global learners who require "big picture" insights, and hands-on learners who struggle with abstractions.

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Background: This text
was used in Spring
2014 for Donald P.
Visco's undergraduate
course in Equilibrium
Thermodynamics
taught within the
Chemical Engineering
Department at The

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University of Akron.

This course was taken by mostly STEM and Honors students. This was the first semester of the text's publication. The class completed Chapters 1-14

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- no tendency for a
species to change
phases or chemical
react •

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that is in mechanical,
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equilibrium - a system
with more than one
phase present that is in
thermal and
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Part I clearly introduces
the laws of
thermodynamics with
applications to pure
fluids. Part II extends
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phase and chemical equilibrium.

Throughout, Matsoukas focuses on topics that link tightly to other key areas of undergraduate chemical engineering, including separations, reactions, and capstone design.

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Al-Zaytoonah
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